

Crystal structure and chemistry of trilithionite- $2M_2$ and polyolithionite- $2M_2$

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Abstract: The crystal chemistry of three Li-bearing mica- $2M_2$ crystals from pegmatites has been studied by chemical analyses and single crystal X-ray diffraction; their belonging to the trilithionite-polyolithionite join is highlighted by the following compositional ranges in atoms per formula unit [based on $O_{12-(x+y)}(OH)_x F_y$]: $3.198 \leq Si \leq 3.538$, $0.462 \leq [^{IV}]Al \leq 0.811$, $1.195 \leq [^{VI}]Al \leq 1.390$, $0.031 \leq (Fe+Mn) \leq 0.072$, $1.522 \leq Li \leq 1.757$, $0.872 \leq K \leq 0.906$, $0.030 \leq Na \leq 0.073$, $0.000 \leq (Cs+Rb) \leq 0.099$, $1.541 \leq F \leq 1.722$. The correlation between F and Li content is confirmed, as observed in Li-rich micas.

Crystal structure refinements were carried out in space group $C2/c$ (R values vary between 0.030 and 0.031). The crystal chemistry is mostly influenced by tetrahedral chemical composition. Increasing $[^{IV}]Al$ content, α and ψ_{MI} parameters increase; Si content involves a lowering of the interlayer separation and tetrahedral thickness. Li content affects octahedral thickness. The stability of $2M_2$ polytype seems to be induced by a relative increase of Δz tetrahedral parameter, which reduces the repulsion between basal tetrahedral oxygen atoms. Unlike Li-bearing muscovite, trioctahedral Li-bearing mica crystals show an octahedral occupancy not related to octahedral charge.

Key-words: mica, trilithionite, polyolithionite, lithium, $2M_2$ polytype.

Introduction

The crystal structure, composition and polytypism of the micas were recently reviewed (Brigatti & Guggenheim, 2002; Ferraris & Ivaldi, 2002; Nespolo & Đurović, 2002). These contributions evidence the significant advancement in describing the crystal chemistry of true micas belonging to $1M$ polytype, mostly showing trioctahedral occupancy, and $2M_1$ polytype, mostly dioctahedral. More limited results are available for $2M_2$, $3T$ and $2O$ sequences both true and brittle micas.

$2M_2$ polytype was detected for a limited set of samples, mostly Li-containing trioctahedral micas (Takeda *et al.*, 1971; Sartori *et al.*, 1973; Guggenheim, 1981) or dioctahedral micas with uncommon interlayer composition (Zhoukhlistov *et al.*, 1973; Ni & Hughes, 1996). Trioctahedral micas- $2M_2$ in the system K-Li-Fe-Al-Si [Takeda *et al.*, 1971; Sartori *et al.*, 1973 and Guggenheim, 1981 ("lepidolite- $2M_2$ " from Radkovice, Czech Republic)] as well as the dioctahedral nanningite- $2M_2$ (Ni & Hughes, 1996) were refined in the space group $C2/c$ and indicate the presence of a large *trans*-site and two equivalent *cis*-sites, whereas ordering of tetrahedral cations is unusual. Unlike $1M$ polytype, no octahedral ordering between octahedral *cis*-sites was indicated for $2M_2$ polytypic arrangement in Li-bearing micas.

At present, there are relatively few published studies on crystals belonging to the $2M_2$ polytype in the K-Li-Fe-Al-Si trioctahedral mica system. Accordingly, this paper attempts to: (1) introduce three new crystal structure refinements of trioctahedral mica- $2M_2$ crystals with unusual composition between trilithionite and polyolithionite; (2) identify the ordering pattern of the octahedral and tetrahedral sites along the trilithionite-polyolithionite join; (3) compare the $2M_1$ and $2M_2$ long-range ordering for crystals along the Al^{3+} - Li^+ join.

Samples

Samples under examination come from Varuträsk (Västerbotten, Sweden) and from Chèdeville (France). The sample from Varuträsk (label: SBT) is from the Varuträsk pegmatite. It is mostly made from mica species, pink in colour, with maximum crystal dimensions of 9 mm. Mica is also associated to albite and stibiotantalite, an uncommon Sb, Ta, and Nb oxide. The outcrop from Varuträsk, genetically related to the Revsunb granite (Quensel, 1955), is a complex granitic LCT pegmatite, according to the Černý nomenclature (Černý, 1992).

Samples from Chèdeville (labels: Lch 59a and Lch 132) were also selected from LCT aplitic pegmatites

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